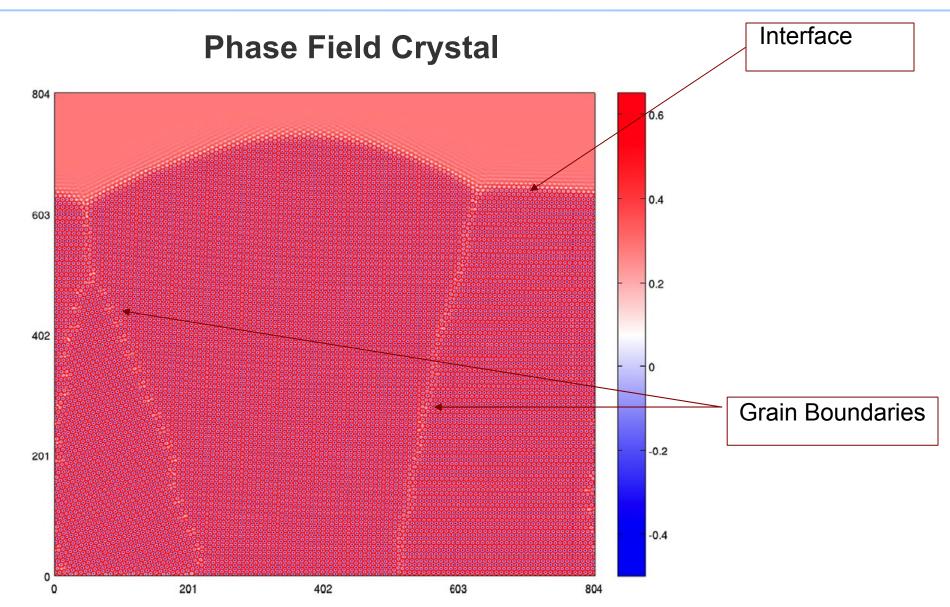


Crystal Model Based on Dynamic Density Functional Theory (DDFT)

Arvind Baskaran and John Lowengrub

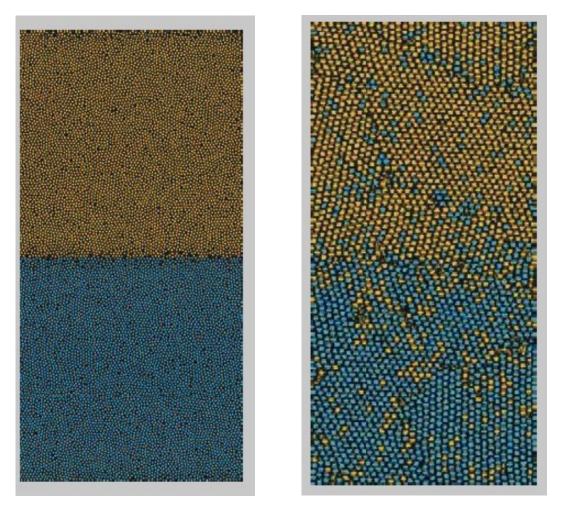
Pyrite Project Meeting December 7th 2010

UNIVERSITY of CALIFORNIA • IRVINE





Phase Field Simulation of Binary Alloy [N Provatas et al JOM (2007)]

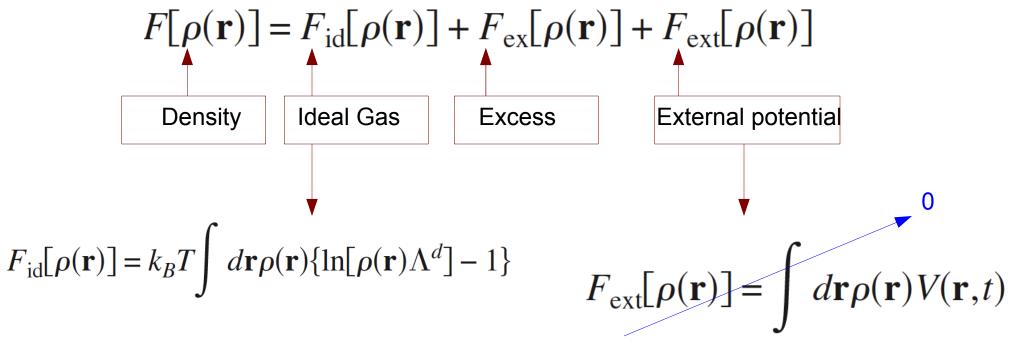




Density functional Theory of Freezing

[Ramakrishanan and Youssouf Phys. Rev. B 19, 2775 (1979)]

The Free Energy of the liquid near freezing point is written as a functional of The scaled atomic density :



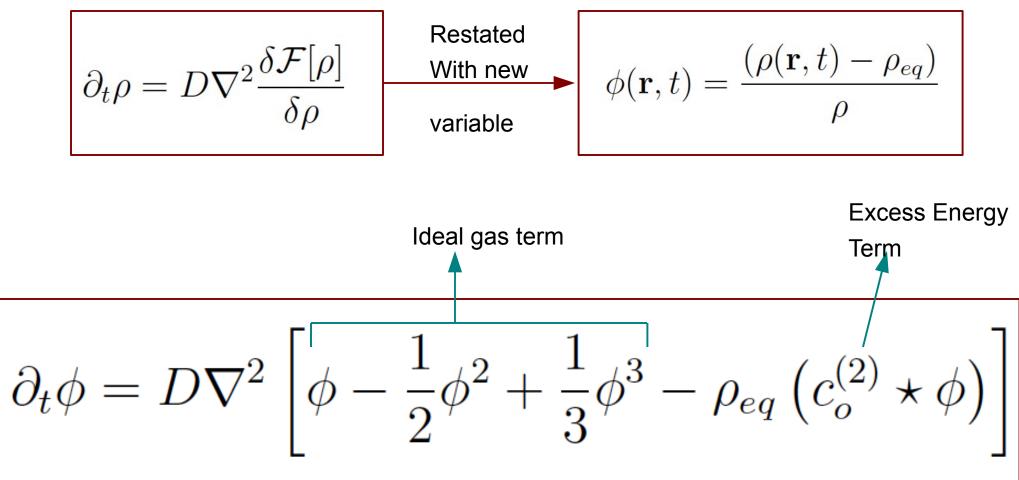


Excess Free Energy Approximation (truncated Taylor Expansion)

The excess Free energy term was expanded by Ramakrishnan and Youssouf in terms of the Density difference $\Delta \rho = \rho(\mathbf{r}) - \rho$ eq Equilibrium density Of liquid $F_{\rm ex}[\rho(\mathbf{r})] \simeq F_{\rm ex}(\rho) - \frac{k_B T}{2} \left[\int d\mathbf{r} d\mathbf{r}' \Delta \rho(\mathbf{r}) \Delta \rho(\mathbf{r}') \times c_0^{(2)}(\mathbf{r} - \mathbf{r}';\rho) \right]$ **Direct Correlation function**



Rescaled DDFT Model





Direct Correlation Function

Ornstein Zernike Relation :

$$h(\mathbf{r}) - c_o^{(2)}(\mathbf{r}) = \rho_{eq} \int h(\mathbf{r}') c_o^{(2)}(\mathbf{r} - \mathbf{r}') d\mathbf{r}'$$

Total correlation function

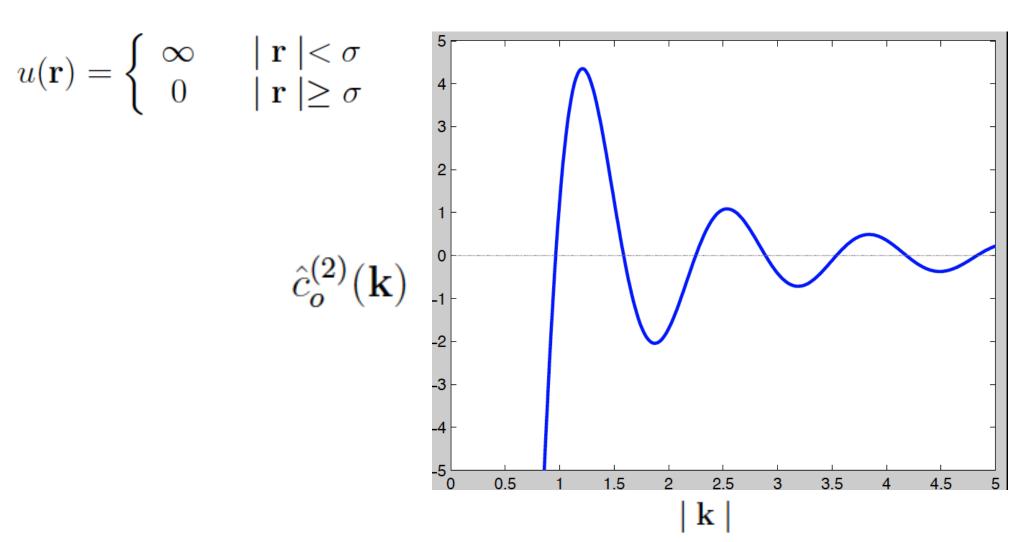
Percus Yevick (PY) Closure Relation :

$$c_o^{(2)} = \left[\exp\left(-\frac{u(\mathbf{r})}{kT}\right) - 1 \right] \left[h(\mathbf{r}) - c_o^{(2)}(\mathbf{r}) + 1 \right]$$



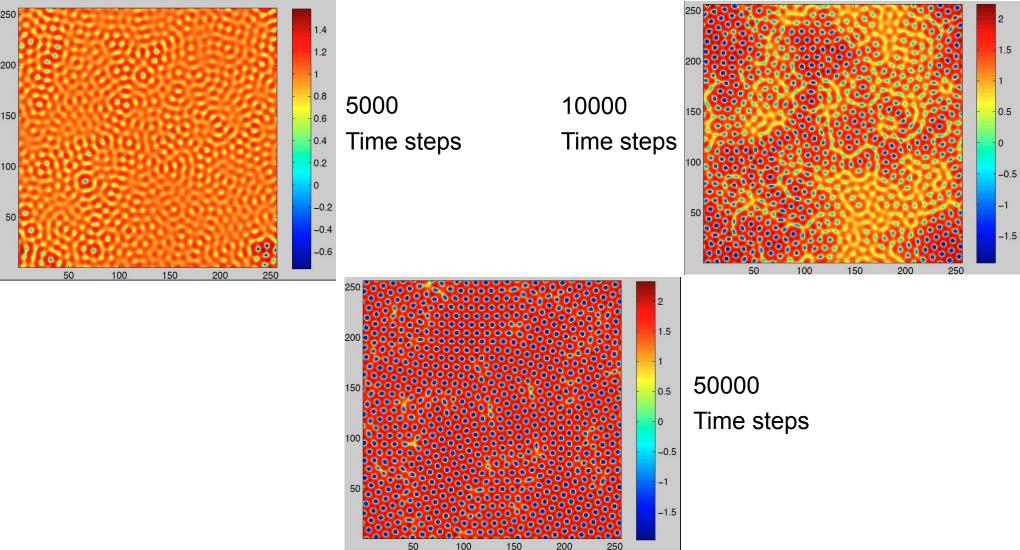


Direct Correlation Function for Hard Sphere with PY Closure



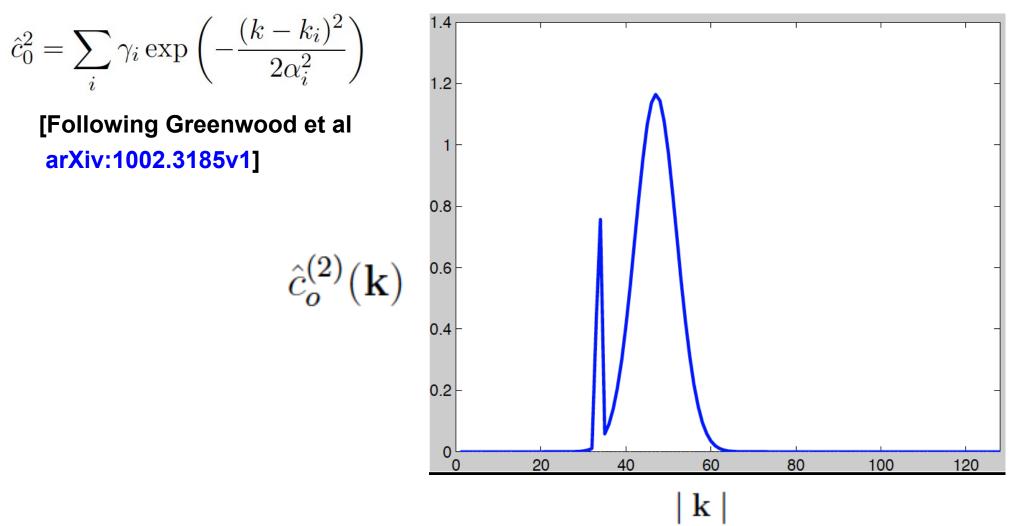


DDFT Simulation of Freezing of a Supercooled Hard Sphere Fluid



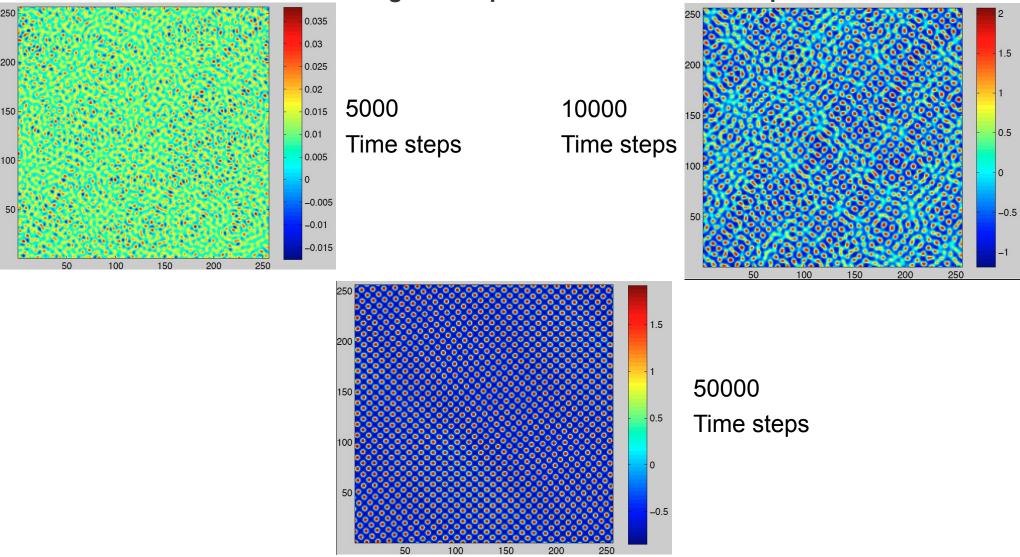


Direct Correlation function for Square Lattice (constructed)





Simulation of Freezing of a Supercooled Fluid into a Square Lattice





Phase Field Crystal Free Energy

The Excess Free Energy is written as :

$$\mathcal{F}_{\text{ex}}[\rho(\mathbf{r})] = F_{\text{ex}}(\rho) - \frac{k_B T}{2} \int d\mathbf{r} \Delta \rho(\mathbf{r})$$

$$\times (\hat{C}_0 - \hat{C}_2 \nabla^2 + \hat{C}_4 \nabla^4 + \cdots) \Delta \rho(\mathbf{r})$$

PFC Approximation to Direct Correlation Function :

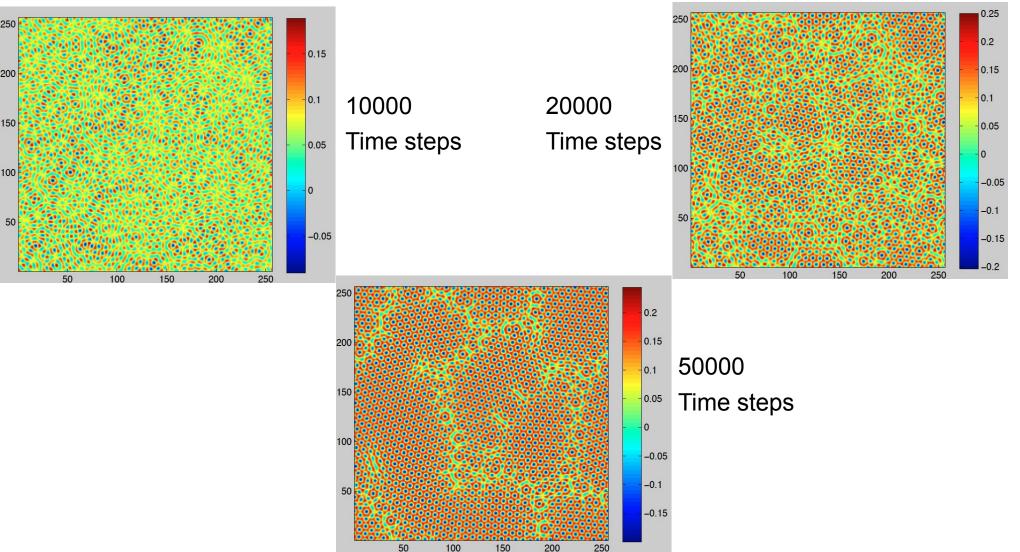
Captures First peak Of Structure Factor

$$\hat{c}_{0}^{(2)}(\mathbf{k};\rho) = \hat{C}_{0} + \hat{C}_{2}k^{2} + \hat{C}_{4}k^{4}$$

$$\widehat{\mathcal{G}}_{0}^{2}$$
Elder et al [PRE 051606 (2004)]



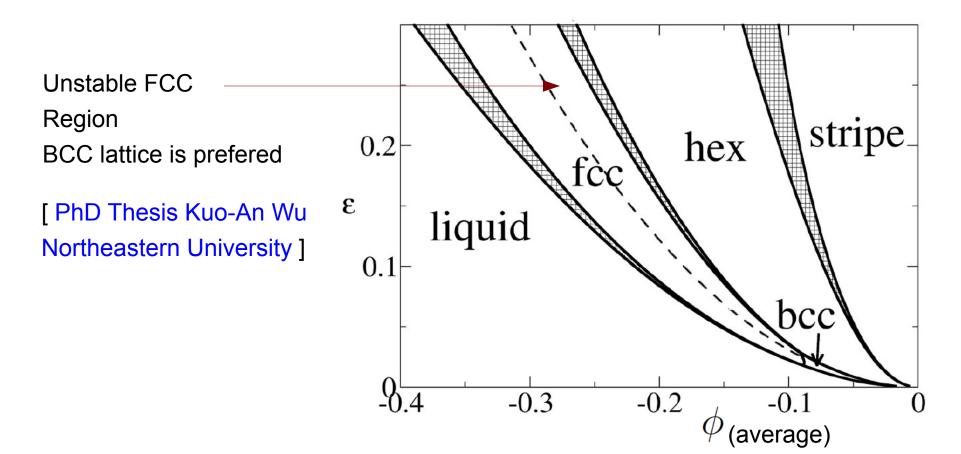
PFC Simulation of Freezing of a Supercooled Fluid





PFC Model in 3 – D (Phase Diagram)

The PFC Model extends directly to 3-D but fails to capture the FCC structure in 3-D

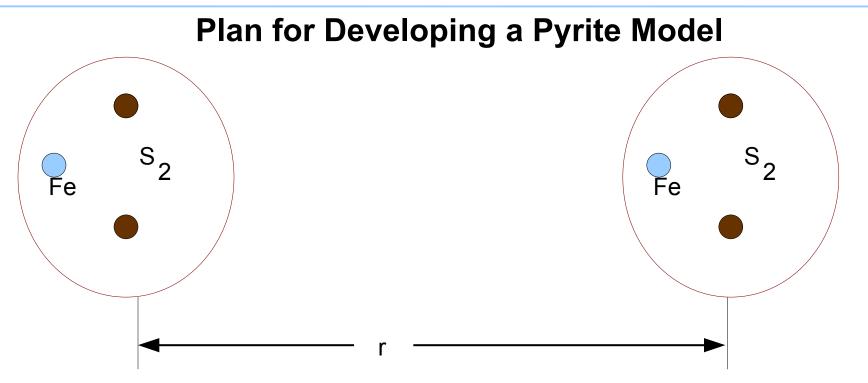




Main Accomplishments Since Last Meeting

- Development of a Algorithm that simulates solidification of a fluid given the Direct correlation function of the fluid phase in 2-D.
 (a spectral method with Crank Nicholson like time integration)
- Development of a Numerical Ornstein Zernike Solver based on the algorithm
 Proposed by Labik et al in [Molecular Physics 1985 (56) 701 -715]
- 3) Development of a 3-D Code for simulation of solidification.





In Collaboration with Ruqian Wu's Group :

Step 1 : Treat Ferrite as a molecule and determine Pair potential using DFT

Step 2 : Develop a model for surface re-arrangements of Sulphur

Step 3 : Add internal degrees of freedom for the Fe S2 molecule (relative orientation etc)



Alternative Approach for Developing a Pyrite Model

- Use Existing Molecular Dynamics Models to obtain pair potentials
 For pairwise interaction of Fe-Fe, S-S and Fe S to model crystallization.
- For simulation of solidification from an ink the treatment of the Solvent Solute interaction is the key ingredient which needs careful modeling.



Relative Comparison of Approaches

DFT Approach:

- The pair potentials are calculated from DFT rather than modeled
- 2) Simpler implementation as theMolecule is assumed to be rigid.But certain defects are hard to capture
- Solute/Solvent interaction (growth from an ink) is not straight forward.

Alternative Approach:

- 1) The pair potentials are modeled
- Molecules not rigid and hence order In the crystal must follow from the Pair interactions.

Thus defects are naturally captured.

Solute/Solvent interaction can be modeled.